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(54) **Rare earth oxide layer on a GaAs- or GaN-based semiconductor body**

(57) We have found that a single crystal, single domain oxide layer of thickness less than 5 nm can be grown on a (100) oriented GaAs-based semiconductor substrate. Similar epitaxial oxide can be grown on GaN and GaN-based semiconductors. The oxide typically is a rare earth oxide of the Mn_2O_3 structure (e.g., Gd_2O_3).

The oxide/semiconductor interface can be of high quality, with low interface state density, and the oxide layer can have low leakage current and high breakdown voltage. The low thickness and high dielectric constant of the oxide layer result in a MOS structure of high capacitance per unit area. Such a structure advantageously forms a GaAs-based MOS-FET.

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Description**Cross Reference to Related Application**

[0001] This application is a continuation-in-part of US patent application Serial No. 09/093,557, filed June 8, 1998 by Hong et al., which is a continuation-in-part of US patent application Serial No. 08/741,010, filed October 31, 1996 by Chen et al., which is a continuation-in-part of US patent application Serial No. 08/408,678, filed March 22, 1995 by Hong et al., which is a continuation-in-part of US patent application Serial No. 08/217,332, filed March 23, 1994, now US patent 5,550,089. See also US patent application Serial No. 08/948,874, filed October 10, 1997 by Hobson et al., which discloses an article comprising an oxide layer on GaN, and a method of making the article.

Technical Field

[0002] This invention pertains to articles that comprise an oxide layer on a GaAs-based or GaN-based semiconductor body, e.g., a field effect transistor (FET).

Background of the Invention

[0003] GaAs-based transistors and circuits are widely used in, for instance, wireless communication apparatus, due inter alia to the relatively high electron mobility in GaAs, the availability of semi-insulating GaAs substrates, and the relative simplicity of manufacturing processes.

[0004] Much effort has been directed towards GaAs-based MOS-FETs. See, for instance, T. Mimura et al., IEEE Transactions on Electron Devices, Vol. ED-27(6), p. 1147 (June 1980) for a review of early work. The authors of that paper concluded (p. 1154) that, although the main features of the results achieved so far are promising, "... some technological problems remain, including anomalous behavior of the dc and low-frequency operation of the devices. Undoubtedly, these problems are associated with the high density of surface states involved in the GaAs MOS system." See also A. Colquhoun et al., IEEE Transactions on Electron Devices, Vol. ED 25(3), p. 375 (March 1978), and H. Takagi et al., IEEE Transactions on Electron Devices, Vol. ED 25 (5), p. 551 (May 1978). The former discloses a device that comprises an etched notch that defines the channel thickness. Such a non-planar structure would be relatively difficult to make repeatably, and thus is less desirable than a planar MOS-FET would be.

[0005] As pointed out by Mimura et al., the early devices suffered from poor gate oxide/GaAs interface quality, including a high density of interface states. In recent years, substantial effort has been directed at this problem.

[0006] For instance, US patent 5,451,548 discloses formation of a Ga₂O₃ film on GaAs by e-beam evapo-

ration from a high purity single crystal of Gd₃Ga₅O₁₂. See also US patent 5,550,089, and US patent application Serial No. 08/408,678, incorporated herein by reference, which disclose GaAs/Ga₂O₃ structures with midgap interface state density below 10¹¹ cm⁻² eV⁻¹. See also M. Passlack et al., Applied Physics Letters, Vol. 69 (3), p. 302 (July 1996) which reports on the thermodynamic and photochemical stability of low interface state density GaAs/Ga₂O₃/SiO₂ structures that were fabricated using in situ molecular beam epitaxy. Other pertinent publications are M. Passlack et al., Applied Physics Letters, Vol. 68(8), p. 1099 (Feb. 1996); and M. Hong et al., J. of Vacuum Science and Technology B, Vol. 14(3), p. 2297, (May/June 1996).

[0007] The parent of this continuation-in-part application discloses GaAs-based semiconductor bodies with an oxide layer thereon, the oxide having overall composition Ga_xA_yO_z, where Ga is substantially in a 3+ oxidation state, A is one or more electropositive stabilizer element adapted for stabilizing Ga in the 3+ oxidation state, x is greater than or equal to zero, z is selected to satisfy the requirement that both Ga and A are substantially fully oxidized, and y/(x+y) is greater than 0.1. Thus, in one embodiment x=0, and A is Gd, the oxide being Gd₂O₃.

[0008] U. S. patent applications Serial No. 09/122,558, filed July 24, 1998 by Chen et al., and Serial No. 09/156719, filed September 18, 1998 by Hong et al., disclose methods of making GaAs-based MOS-FETs having gate oxide of composition Ga_xA_yO_z, as described above.

[0009] A MOS structure basically is a planar capacitor, with the capacitance depending linearly on the dielectric constant of the oxide, and inversely on the thickness of the oxide layer. In order to obtain GaAs-based MOS structures with increased capacitance per unit area, it would be desirable to have available structures with oxide layer thickness less than previously required to give acceptably low leakage current. This application discloses such GaAs-based MOS structures.

[0010] All patents, patent applications, and scientific papers that are cited herein are incorporated into this patent application by reference.

Summary of the Invention

[0011] As will be described in more detail below, the parent of this continuation-in-part application discloses articles comprising a novel dielectric layer on GaAs-based semiconductors, and a method of making the article.

[0012] More specifically, the article comprises an oxide layer on a GaAs-based semiconductor body, and forming an interface therewith. The article further comprises a metal contact disposed on each of the oxide layer and the semiconductor body. Associated with the interface is a midgap interface state density of at most 1 x 10¹¹ cm⁻² eV⁻¹ at 20°C.

[0013] Significantly, the oxide layer has overall composition $Ga_xA_yO_z$, where Ga substantially is in a 3+ oxidation state, where A is one or more electropositive stabilizer element for stabilizing Ga in the 3+ oxidation state, x is greater than or equal to zero, $y/(x+y)$ is greater than or equal to 0.1, and z is sufficient to satisfy the requirement that Ga and A are substantially fully oxidized. Herein, Ga and A each are considered to be "substantially fully oxidized" if at least 80% (preferably at least 90%) of the respective element is fully oxidized, i.e., is in the highest oxidation state of the element. The highest oxidation state of Ga is 3+. The highest oxidation state of A depends on A. For instance, if A is an alkaline earth, then the state is 2+, and if A is Sc, Y, or a rare earth element, then the state is frequently, but not always, 3+.

[0014] The method of making the article comprises the steps of providing the GaAs-based semiconductor body, treating the body such that at least a portion of a major surface of the body is essentially atomically clean and essentially atomically ordered, forming, substantially without exposure of the semiconductor body to contamination, the oxide layer on the essentially atomically clean and ordered surface, and forming the metal contacts.

[0015] Significantly, the first forming step comprises forming the oxide layer such that the oxide layer has overall composition $Ga_xA_yO_z$, where Ga substantially is in a 3+ ionization state, where A is one or more electropositive stabilizer element for stabilizing Ga in the 3+ ionization state, x is greater than or equal to zero, $y/(x+y)$ is greater than or equal to 0.1, and z is sufficient to satisfy the requirement that Ga and A are substantially fully oxidized.

[0016] In an exemplary embodiment of the article, the oxide contains both Ga and A, and the stabilizer element A is Sc, Y or a rare earth (atomic number 57-71) element. In another exemplary embodiment the oxide layer is an essentially Ga-free oxide of a stabilizer element.

[0017] In an exemplary embodiment of the method the oxide layer is formed by simultaneous deposition from two (or possibly more) deposition sources, with one of the sources containing Ga_2O_3 (typically in powder form), and the other containing an oxide of a stabilizer element (e.g., Gd_2O_3), typically also in powder form. In another exemplary embodiment the oxide layer is formed by deposition from a single deposition source containing an oxide of a stabilizer element, e.g., Gd_2O_3 .

[0018] We have now made the surprising discovery that Gd_2O_3 can be grown epitaxially, in single crystal, single domain form, on a GaAs-based semiconductor substrate. The single crystal oxide layer has low leakage current even for very small layer thickness, exemplarily less than 5 nm, and is advantageously used as gate oxide layer. By "GaAs-based semiconductor" we mean GaAs, ternary III/V alloys such as InGaAs, and quaternary Ga and As-containing III/V alloys such as AlInGaAs. Furthermore, we have found that Gd_2O_3 can be grown in single crystal, single domain form on GaN and

GaN-based semiconductors.

[0019] The above-described discovery makes possible the fabrication of MOS structures of increased capacitance per unit area. The increased capacitance is primarily due to the ability to use a thinner oxide layer than was previously possible.

[0020] We currently believe that the invention is not limited to the use of Gd_2O_3 but that, in view of the known similarity to each other of the rare earth elements, other rare earth oxides may also be used, provided the oxides have cubic structure of the Mn_2O_3 type.

[0021] More specifically, our invention is embodied in an article that comprises a planar single crystal semiconductor substrate with an oxide layer on the semiconductor substrate.

[0022] Significantly, the semiconductor substrate is selected from the group consisting of the GaAs-based III/V semiconductors, and the oxide is selected from the group consisting of the oxides that have a cubic structure of the Mn_2O_3 type and composition X_2O_3 , where X is a rare earth element or Y, exemplarily, one or more of Y, La, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb and Lu. The oxide layer is a single crystal, single domain layer, preferably of thickness less than 5 nm. By a "single domain" single crystal layer we mean herein a crystal layer having identically the same crystal orientation throughout.

Brief Description of the Drawing's

[0023]

FIG. 1 schematically depicts an exemplary device, namely, a MOS-FET;

FIG. 2 shows the diagram of an exemplary circuit comprising complementary MOS-FETs;

FIGs. 3-5 show electrical characteristics of an exemplary enhancement-mode p-channel GaAs MOS-FET;

FIGs. 6-8 show measurement data from GaAs/oxide/metal structures of various oxide compositions; FIGs. 9-11 show in-situ RHEED patterns of, respectively, a (100) GaAs surface along [011] and $[01\bar{1}]$ axes, a 1.5 nm thick (110) Gd_2O_3 film along [001], $[\bar{1}10]$ and $[\bar{1}11]$ axes; and a 18.5 nm thick (110) Gd_2O_3 film along the same three axes;

FIGs. 12-13 show, respectively, x-ray $\theta - 2\theta$ scans of the (440) Bragg reflection of the (110) oriented Gd_2O_3 films of 18.5 nm, 3.5 nm and 1.5 nm, and rocking curve scans of these peaks about the θ angle for the three samples;

FIG. 14 shows the leakage current density versus applied field for Gd_2O_3 films with thicknesses of 26.0, 18.5, 14.0, 3.5 and 1.5 nm thickness, respectively.

[0024] Like or analogous features generally are designated by the same numeral in different figures. The

figures are not to scale.

Detailed Description of the Invention

[0025] To the best of our knowledge, prior art GaAs-based MOS structures comprise an amorphous oxide layer. Indeed, we are unaware of any suggestion in the prior art that such a structure could be formed with a single crystal oxide layer.

[0026] FIG. 1 schematically depicts an exemplary electronic device, namely, a GaAs-based MOS-FET. Numerals 140-147 refer, respectively, to the GaAs body (exemplarily p-type), the source region (exemplarily n-type), the drain region (exemplarily n-type), the drain contact, the source contact, the gate oxide, the gate contact and the field oxide. Associated with the interface between 140 and 145 is an interface state density $< 10^{11}/\text{cm}^2\cdot\text{eV}$, and typically a recombination velocity $< 10^4\text{cm/s}$.

[0027] The parent of this continuation-in-part application discloses in detail a process of making a GaAs-based MOS-FET structure. The process can be used to make such a structure with oxide layer of overall composition $\text{Ga}_x\text{Al}_y\text{O}_z$, where $x \geq 0$, and the other symbols are as defined above. The parent application also discloses that the oxide layer thickness will typically be in the approximate range 5-150 nm. An important aspect of the process is the condition of the substrate surface. The surface has to be essentially atomically clean and essentially atomically ordered, i.e., the surface is "reconstructed".

[0028] The thus produced MOS-FET desirably will be electrically connected in conventional manner to other electronic devices, including other MOS-FETs, to form an integrated circuit. By way of example, complementary MOS-FETs can be connected as shown in FIG. 2 to provide an inverter. In FIG. 2, numeral 231 refers to a n-channel enhancement mode GaAs MOS-FET and 232 refers to a p-channel enhancement mode GaAs MOS-FET. The circuit per se is conventional, but the implementation in GaAs MOS-FET technology is novel, to the best of our knowledge. The combination of FIG. 2 is representative of circuits according to the invention.

[0029] As those skilled in the art will recognize, the described process not only is suitable for producing GaAs-based planar n-channel and p-channel MOS-FETs on the same substrate, but can also be used to combine such MOS-FETs with GaAs (metal-semiconductor) MES-FETs. In order to produce a MES-FET, the gate oxide exemplarily is removed from the respective gate region, and an appropriate metal (e.g., Ti/Pt/Au) is deposited in the gate region to provide a Schottky barrier contact. GaAs-based MES-FETs are known and do not require detailed description. However, the ability to integrate GaAs MES-FETs with GaAs MOS-FETs (including complementary enhancement type GaAs-MOS-FETs) does, to the best of our knowledge, not exist in the prior art.

[0030] GaAs-based MOS-FETs were produced substantially as described in the parent of this continuation-in-part application. A thus produced device ($40 \times 50\mu\text{m}^2$ gate geometry) was tested, and found to have the following characteristics:

[0031] As demonstrated by FIG. 3, the gate oxide had breakdown voltage of $3.6 \times 10^6 \text{ Vol/cm}$. The transconductance was 0.3 mS/mm . FIG. 4 shows the drain I-V characteristics of the exemplary MOS-FET, and FIG. 5 shows drain current vs. gate voltage.

[0032] It is apparent from a variety of references (e.g., US patents 5,550,089 and 5,597,768; M. Passlack et al., *Applied Physics Letters*, Vol. 69(3), pp. 302-304) that it is widely assumed in the art that e-beam evaporation from single crystal GGG (gadolinium gallium garnet; $\text{Gd}_3\text{Ga}_5\text{O}_{12}$) can result in the deposition of a thin film that is almost pure Ga_2O_3 , with only a small amount (e.g., 0.1 at %) Gd also present in the film.

[0033] Furthermore, it is believed that the presence of the Gd is undesirable, and that ideally the film should be pure Ga oxide. See, for instance, US patent 5,597,768, which discloses at col. 1, lines 39-45 that "... The major problem is that some of the Gd_2O_3 (about 0.1% according to the patent) is incorporated into the thin film as impurities creating defects increasing the bulk trap density. The increased defects and, in turn, increased bulk trap density degrades the performance of the devices in which the thin film is used." In order to decrease the "impurity" level in the Ga_2O_3 thin film, the '768 patent teaches use of different evaporation source materials that "contain" Ga_2O_3 and another oxide having a melting temperature that is more than 700°C above the sublimation temperature of Ga_2O_3 . One such material is MgGa_2O_4 , which is said to "contain" MgO and Ga_2O_3 . The use of MgGa_2O_4 instead of GGG is said to result in "... drastically reduced incorporation of the undesired specie in the oxide film and in significantly lower bulk trap density in the oxide film." See col. 3, lines 18-21 of the '768 patent. However, the '768 patent does not provide any experimental data.

[0034] Our continued research (which included Rutherford Backscattering Spectrometry (RBS) and Auger analysis of very thin deposited films) now has yielded results that lead to a different approach to making a device-grade (e.g., interface state density in the $10^{10}\text{cm}^{-2}\text{eV}^{-1}$ range, specifically, $1 \times 10^{11}\text{cm}^{-2} \text{eV}^{-1}$ or below; with low leakage current and high breakdown voltage) thin oxide film on GaAs and GaAs-based semiconductor bodies.

[0035] We have found inter alia that, contrary to the teaching of the '768 patent, the oxide thin film desirably contains a substantial amount of Gd (or other appropriate metal element). Indeed, we have found that a film that is formed by evaporation from pure Ga_2O_3 powder (and thus is essentially pure Ga oxide) is generally not of device quality, whereas a film that is formed by evaporation from pure Gd_2O_3 (and thus is essentially pure Gd oxide) is generally of device quality, with low inter-

face state density. This is exemplified by FIG. 6, which shows data on the I-V (current-voltage) characteristics of very thin oxide layers (16.5 nm - 40 nm) on GaAs, with a conventional metal contact defining the area of the MOS capacitor. As can be readily seen from FIG. 6, a 40 nm thick Ga_2O_3 film has essentially zero breakdown voltage for negative bias, with relatively high leakage for positive bias, and thus is unsuitable for most device applications, whereas a 18.5 nm thick Gd_2O_3 film conducts a mere 10^{-8} A/cm² at + or -3V. Films of intermediate composition (Gd content 6, 14 and 20% of the total metal content of the film) show intermediate breakdown voltages.

[0036] We have found that a thin oxide film that is formed by in situ coevaporation from two separate sources, e.g., one using Ga_2O_3 powder and the other Gd_2O_3 powder, generally will be of device quality, provided the evaporation rates are selected such that the content of Gd (or other appropriate metal element) is at least about 10 atomic % of the total metal content, preferably at least 20 atomic %.

[0037] It will be understood that the GaAs substrates were prepared as described above (resulting in an essentially atomically clean and atomically ordered surface), with the evaporations carried out in situ, without exposure of the samples to the atmosphere.

[0038] The above described experimental facts are consistent with the requirement that in mixed oxide films (of overall composition $\text{Ga}_x\text{A}_y\text{O}_z$, with A, x, y and z as defined below) the Ga ions are caused to be in their 3+ oxidation state. This is facilitated through appropriate choice of stabilizer element and oxygen content.

[0039] It is known that Ga can exist in the 1+, 2+ and 3+ oxidation states, and that there are electro-positive elements (to be termed "stabilizers") that can stabilize Ga in the 3+ state. Among the stabilizers for Ga is Gd, having Pauling electronegativity 1.1. Among other possible stabilizers for Ga are Sc, Y, the other rare earths, the alkaline earths and the alkalis. In addition to provision of a stabilizer element for Ga, it is also required that sufficient oxygen be present in the deposited film to satisfy the requirement that Ga and the stabilizer element are substantially fully oxidized.

[0040] The above exposition is for tutorial purposes only, and is not intended to limit the scope of the invention.

[0041] In exemplary embodiments of the invention of the parent application, oxide film is formed by deposition from 2 (or more) separate deposition sources (e.g., e-beam evaporation sources), with the deposition parameters selected such that the resulting film has overall composition $\text{Ga}_x\text{A}_y\text{O}_z$, where A is one or more electro-positive stabilizer element for stabilizing Ga in the 3+ oxidation state, and $x > 0$. Furthermore, $y/(x+y)$ is greater than or equal to 0.1, preferably ≥ 0.2 , and z is selected to satisfy the requirement that both Ga and A are substantially fully oxidized.

[0042] In other exemplary embodiments, the oxide

film is formed by deposition of A-oxide, without deposition of Ga-oxide. The film thus has overall composition AO_z , with z selected such that A is substantially fully oxidized.

[0043] By way of example (for both $x=0$ and $x>0$), A is selected from the group consisting of Sc, Y, the rare earth metals (atomic number 57-71), the alkaline earths (e.g., Be, Mg, Ca, Sr, Ba), and the alkalis (e.g., Li, Na, K, Rb, Cs).

[0044] The alkali elements generally are not preferred, due to their relative instability to moisture and incompatibility with semiconductor device processing conditions. The currently preferred stabilizer elements are Sc, Y the rare earths (atomic number 57-71) and the alkaline earths, with Sc, Y, La, Nd, Gd, Dy, Ho, Er and Lu being currently most preferred.

[0045] It will be understood that the requirement that, in mixed oxide films, Ga (and A) are substantially fully oxidized does not mean that 100% of all Ga ions have to be in the 3+ ionization state. Typically acceptable results are obtained if 80% or more of all Ga is in the 3+ state, with 80% or more of A also being fully oxidized.

[0046] FIG. 7 shows C-V (capacitance-voltage) data for a 19 nm thick film of Gd_2O_3 on n-type (2×10^{18} cm⁻³) GaAs of (100) orientation. The area of the MOS capacitor was 4.4×10^{-5} cm². FIG. 7 clearly shows accumulation and inversion at low frequency. The data establishes the suitability of the oxide film for device applications, including for MOS-FETs.

[0047] FIG. 8 shows similar data for a 32 nm thick Ga- and Gd-containing mixed oxide film, with about 22% of the total metal content being Gd. The data also shows accumulation and inversion, substantially as the data of FIG. 7.

[0048] In an exemplary preferred embodiment the article comprises a planar enhancement mode MOS-FET with inversion channel.

Example 1

[0049] A MOS-FET is fabricated substantially as disclosed in the parent of this continuation-in-part application. A Gd- and Ga-containing oxide film is deposited by simultaneous e-beam evaporation from a source containing Ga_2O_3 powder and a source containing Gd_2O_3 powder, with the evaporation rates selected to yield a film having Gd - 22% of the total metal content. The MOS-FET has characteristics substantially as shown in FIGs. 3-5.

Example 2

[0050] A MOS-FET is fabricated substantially as in Example 1, except that the oxide film is Gd_2O_3 . The MOS-FET performs substantially as the device of Example 1.

[0051] Although the GaAs-based MOS structures of the parent of this continuation-in-part application exhibit

excellent properties, it would be desirable if further improvement could be obtained. We have succeeded in this, through our surprising discovery that Gd_2O_3 (and also other cubic oxides having the Mg_2O_3 structure) can be deposited on (100) GaAs in epitaxial single crystal, single domain form, with very thin (e.g., less than 5 nm) layers having low leakage current and high breakdown voltage. This is demonstrated by FIG. 14, which shows that an epitaxial Gd_2O_3 film of thickness 1.5 nm has leakage current (at zero voltage) of about 10^{-9} A/cm². The capacitance per unit area of a MOS structure being inversely proportional to the thickness of the dielectric layer, it is evident that a MOS structure according to the invention, with oxide thickness of, e.g., less than 5 nm has significantly greater capacitance per unit area than an analogous prior art structure with substantially greater oxide thickness (e.g., 18.5 nm).

[0052] The growth of the Gd_2O_3 layers on reconstructed (100) GaAs was performed in a multi-chamber UHV system under conditions similar to the evaporation of single crystal $\text{Gd}_5\text{Gd}_3\text{O}_{12}$, as previously described. See, for instance, M. Hong, *J. of Crystal Growth*, Vol. 150, pp. 277-284, 1995, and the parent of this continuation-in-part application.

[0053] Whereas the mixed oxide films formed by evaporation of $\text{Ga}_5\text{Gd}_3\text{O}_{12}$ were amorphous, the films formed by evaporation of Gd_2O_3 (typically in form of packed powder) unambiguously are single crystal, epitaxial with the single crystal (100) GaAs substrate. The oxide is grown in (110) orientation on the GaAs (100) surface.

[0054] Using in-situ reflection high-energy electron diffraction (RHEED) and single-crystal X-ray diffraction, we have determined that the growth of Gd_2O_3 on GaAs is single domain, essentially without dislocations. The in-plane epitaxial relationship was observed for layer thicknesses from 1.5 nm to 40 nm. FIG. 9 shows the (2x4) reconstructed RHEED pattern for an As-stabilized (100) GaAs surface in [011] and $[0\bar{1}\bar{1}]$ directions. Deposition of a 1.5 nm Gd_2O_3 film onto the reconstructed GaAs surface resulted in streaky RHEED patterns of two-fold symmetry, as shown in FIG. 10. The figure also shows the RHEED pattern in the $[\bar{1}\bar{1}11]$ direction. FIG. 11 shows the patterns in the same directions, for an 18.5 nm thick Gd_2O_3 film. The two-fold symmetry of the RHEED pattern shows that the (110) Gd_2O_3 film is grown in single domain on (100) GaAs. The in-plane epitaxial relationship between GaAs substrate and Gd_2O_3 film is as follows: [001] Gd_2O_3 is parallel to [011] GaAs, and $[\bar{1}\bar{1}10]$ Gd_2O_3 is parallel to $[0\bar{1}\bar{1}]$ GaAs.

[0055] FIG. 12 shows x-ray diffraction patterns along the surface normal (110) of Gd_2O_3 films of 18.5, 3.5 and 1.5 nm, respectively. Rocking scans of these peaks about the θ angle are shown in FIG. 13. These and other scans established that the oxide layer has the Mn_2O_3 structure. The thinnest films (e.g., 1.5 nm) show a very sharp component in the rocking scan, indicative of elastic distortion of the lattice, in order to conform to the in-

plane perfect epitaxial condition. As the film thickness grows (e.g., 5 nm or more) it becomes energetically more favorable to relax the strains by generating misfit dislocations.

[0056] Our x-ray work established that, under the above-described conditions, Gd_2O_3 can be formed on (100) GaAs in single crystal, single domain form. This is unusual, considering the two-fold degeneracy of aligning the (110) Gd_2O_3 plane of rectangular symmetry onto the square symmetric (100) GaAs face. The attainment of a single domain may be attributed to the (2x4) reconstruction occurring on the GaAs surface that removes the two-fold degeneracy and favors the single variant growth. In addition, we also found epitaxial growth of a single-domain Gd_2O_3 on a Ga-stabilized (4x6) reconstructed GaAs surface.

[0057] The single crystal Gd_2O_3 films are highly electrically insulating, showing very low leakage current densities in the 10^{-9} to 10^{-10} A/cm² range at zero bias. FIG. 14 shows the dependence of leakage current density on the applied field for a set of Gd_2O_3 samples, with oxide thicknesses in the range from 26 to 1.5 nm. Positive voltage bias corresponds to the top electrode being positive with respect to the GaAs substrate. The breakdown field increases systematically from 3 to 13 MV/cm, and the leakage current density increases only by an order of magnitude as the Gd_2O_3 thickness is decreased from 26 nm to 3.5 nm.

[0058] We also found that Gd_2O_3 films that were grown at substrate temperatures above about 600°C showed high leakage current and low breakdown field. However, once a single crystal Gd_2O_3 film is formed on a GaAs surface at a growth temperature in the approximate range 200°C - 550°C, the film is thermodynamically stable when subsequently subjected to high temperature (e.g., 850°C) processing. This thermodynamic stability facilitates device processing, as those skilled in the art will recognize.

[0059] The above discussion generally is in terms of an epitaxial layer of Gd_2O_3 on either a GaAs-based semiconductor or GaN-based semiconductors. However, it is well known that the rare earth elements all have very similar chemical properties. Thus, we believe that a routine survey will establish that Y, as well as rare earth oxides other than Gd_2O_3 , can also form epitaxially on a GaAs-based semiconductor or GaN. In particular, applicants expect that Y and rare earths that form oxides of composition X_2O_3 and have crystal structure of the Mn_2O_3 type can form epitaxially on a GaAs-based semiconductor or GaN. The rare earths La, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb and Lu are preferred, as is Y.

[0060] The ability to grow single crystal, single domain layers of Gd_2O_3 (as well as oxides of Y and a variety of rare earths) on single crystal GaAs-based III/V semiconductors or GaN-based semiconductors (e.g., AlGaIn or InGaIn), makes possible the epitaxial growth of more complex structures, and all such structures are contemplated. For instance, an epitaxial GaAs/ Gd_2O_3 /GaAs

structure can be readily obtained, and can be advantageously used in semiconductor device manufacture.

Claims

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1. Article comprising a planar single crystal semiconductor substrate having an orientation, and an oxide layer on the semiconductor substrate;

CHARACTERIZED IN THAT

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a) said semiconductor is selected from the group consisting of the GaAs-based semiconductors and GaN-based semiconductors;

b) said oxide layer is selected from the group consisting of the oxides having cubic crystal structure of the Mn_2O_3 type and having composition X_2O_3 , where X is selected from the group consisting of Y and the rare earth elements; and

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c) the oxide layer is a single crystal, single domain oxide layer that is epitaxial with the single crystal semiconductor substrate and has a thickness less than 5nm.

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2. Article according to claim 1, wherein the semiconductor substrate is a (100)-oriented GaAs substrate.

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3. Article according to claim 1, wherein the oxide layer is selected from the group consisting of the oxides of Y, Sc, La, Nd, Gd, Dy, Ho, Er and Lu.

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4. Article according to claim 2, wherein the oxide is Gd_2O_3 .

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5. Article according to claim 1, wherein the article comprises a field effect transistor, with the oxide layer being a gate oxide, of the field effect transistor.

6. Article according to claim 1, further comprising an epitaxial single crystal semiconductor layer disposed on the oxide layer.

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FIG. 1

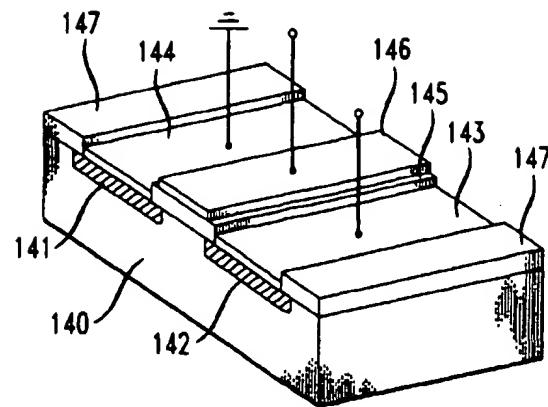


FIG. 2

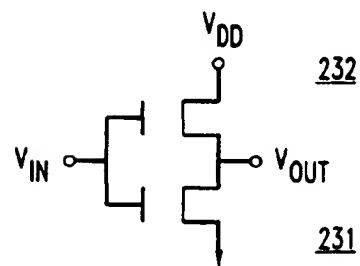


FIG. 3

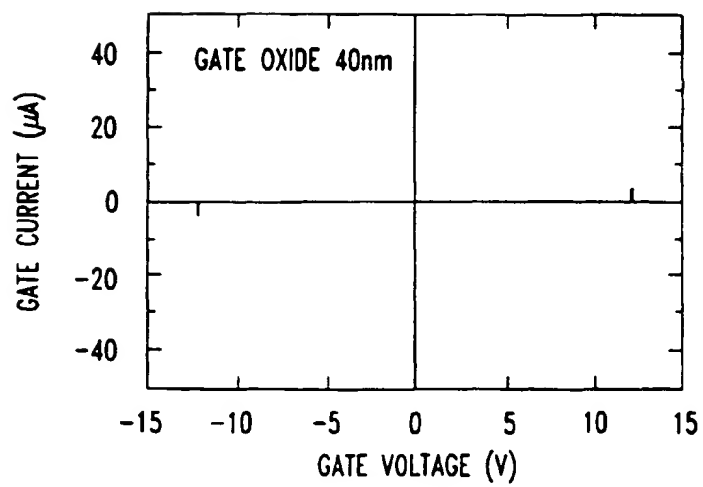


FIG. 4

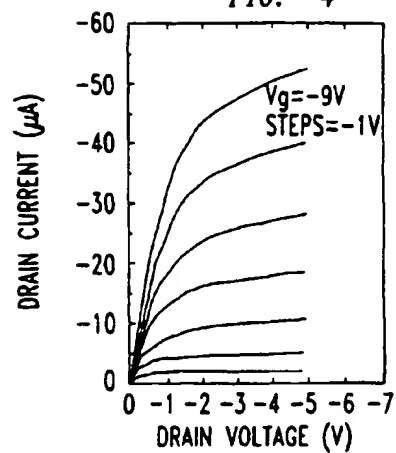


FIG. 5

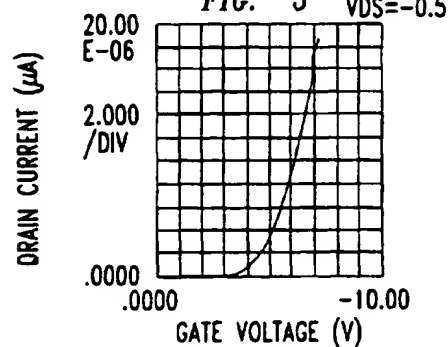


FIG. 6

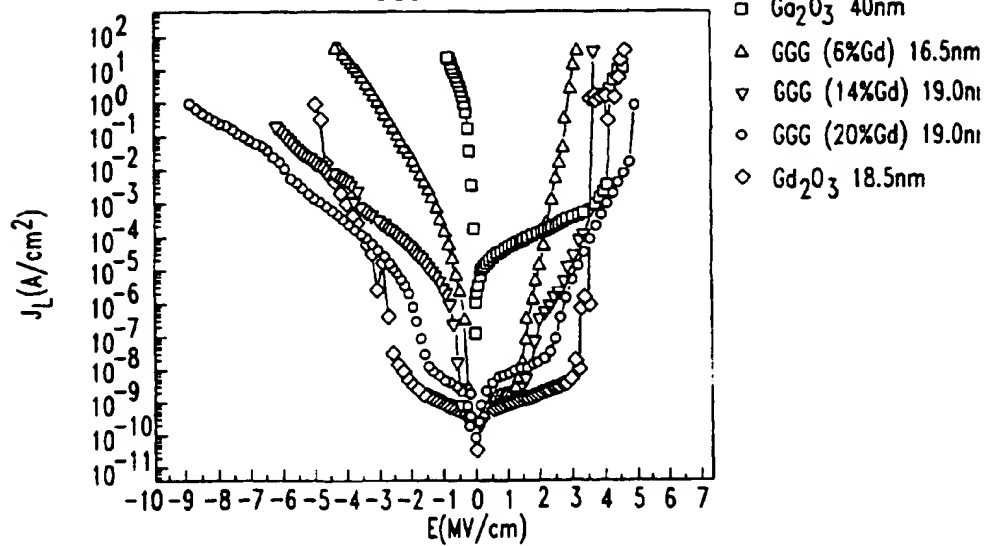


FIG. 7

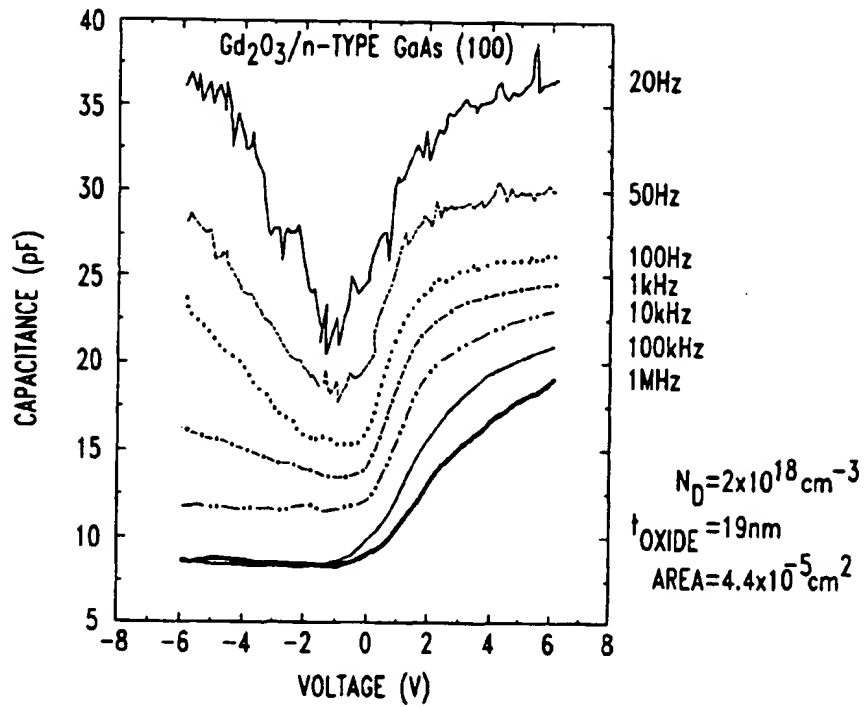


FIG. 8

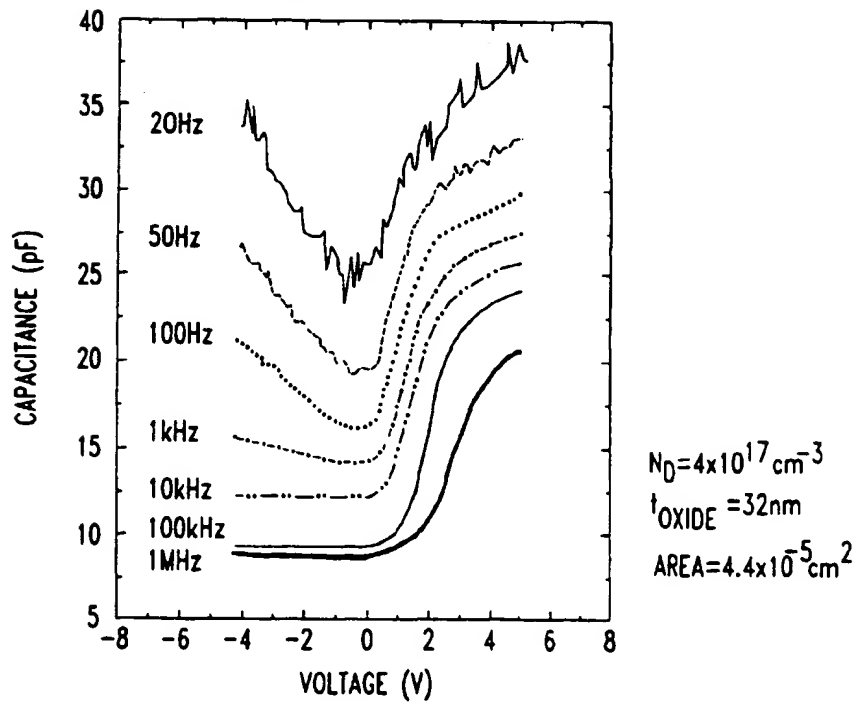


FIG. 9

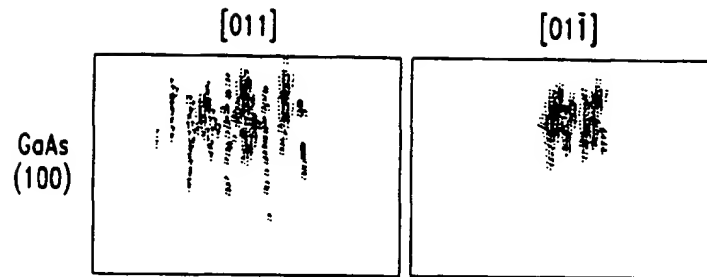


FIG. 10

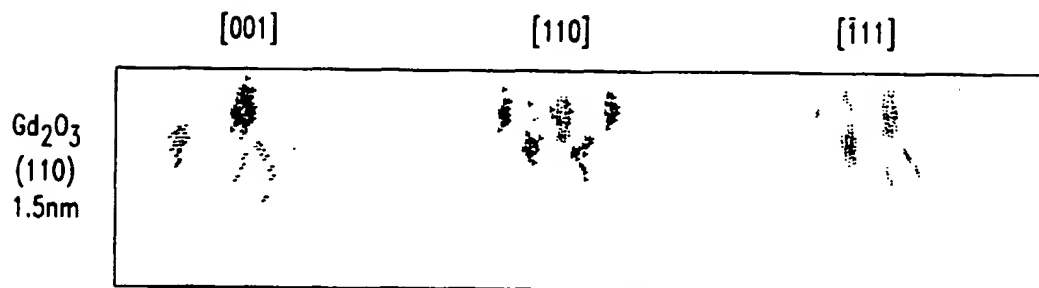


FIG. 11

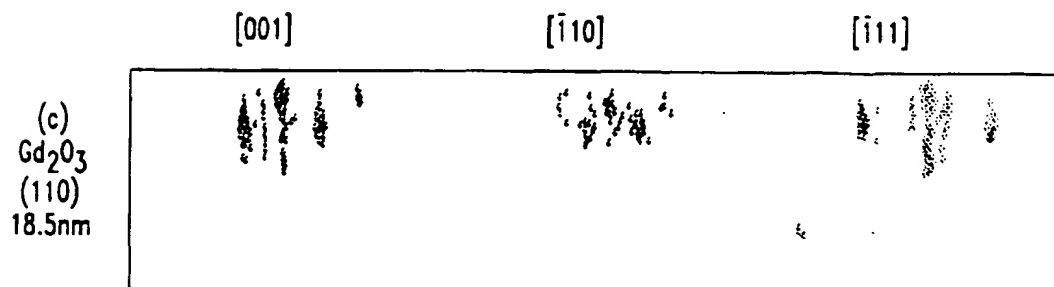


FIG. 12

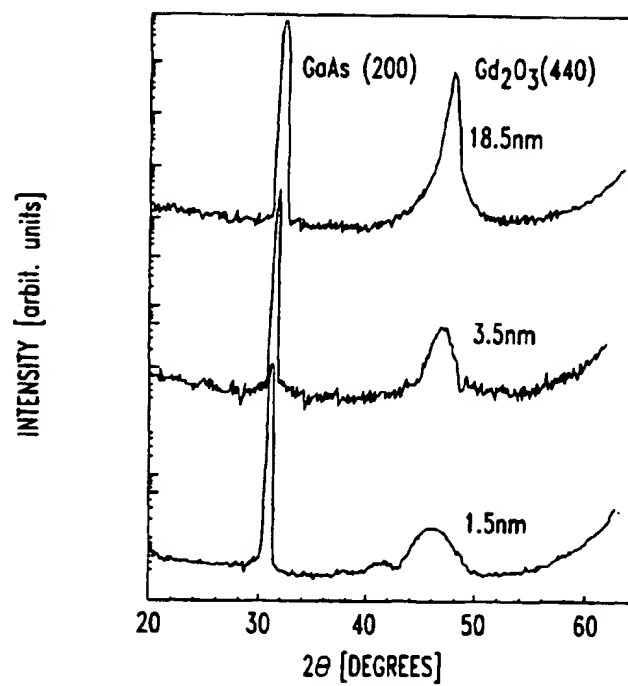


FIG. 13

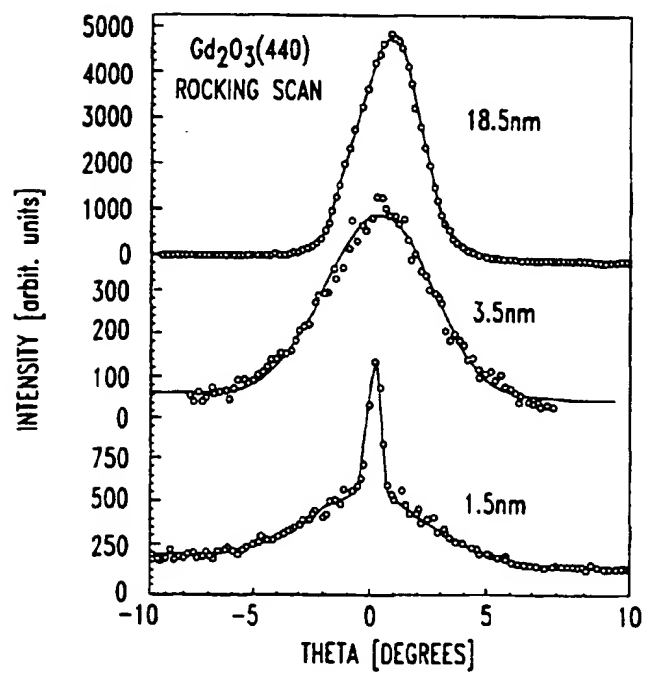
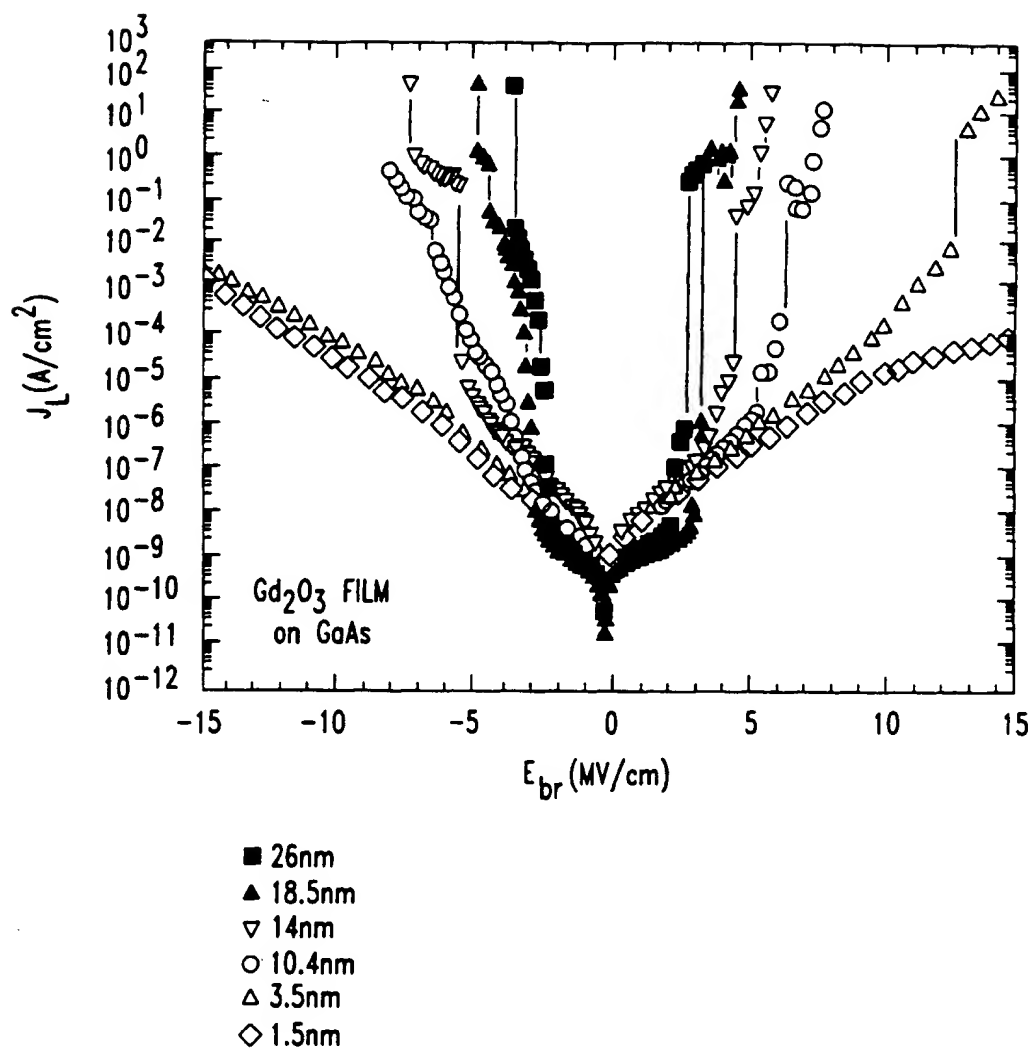


FIG. 14





European Patent
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EUROPEAN SEARCH REPORT

Application Number
EP 99 30 8690

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 0 203 578 A (HITACHI LTD) 3 December 1986 (1986-12-03) * page 5, line 5 - page 6, line 21 * * examples 3,4,7 * * page 12, line 17 - page 13, line 14 * * claims 1,2 *	1-6	H01L29/51
P,X	HONG M ET AL: "Epitaxial cubic gadolinium oxide as a dielectric for gallium arsenide passivation" SCIENCE, vol. 283, no. 5409, 19 March 1999 (1999-03-19), pages 1897-1900, XP002132200 ISSN: 0036-8075 * the whole document *	1-5	
P,X	KORTAN A R ET AL: "Structure of epitaxial Gd2O3 films grown on GaAs(100)" PHYSICAL REVIEW B (CONDENSED MATTER), vol. 60, no. 15, 15 October 1999 (1999-10-15), pages 10913-10918, XP002132201 ISSN: 0163-1829 * the whole document *	1-4	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			H01L C30B
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 7 March 2000	Examiner Köpf, C
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X: particularly relevant if taken alone Y: particularly relevant if combined with another document of the same category A: technological background O: non-written disclosure P: intermediate document</p> <p>T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons &: member of the same patent family, corresponding document</p>			

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